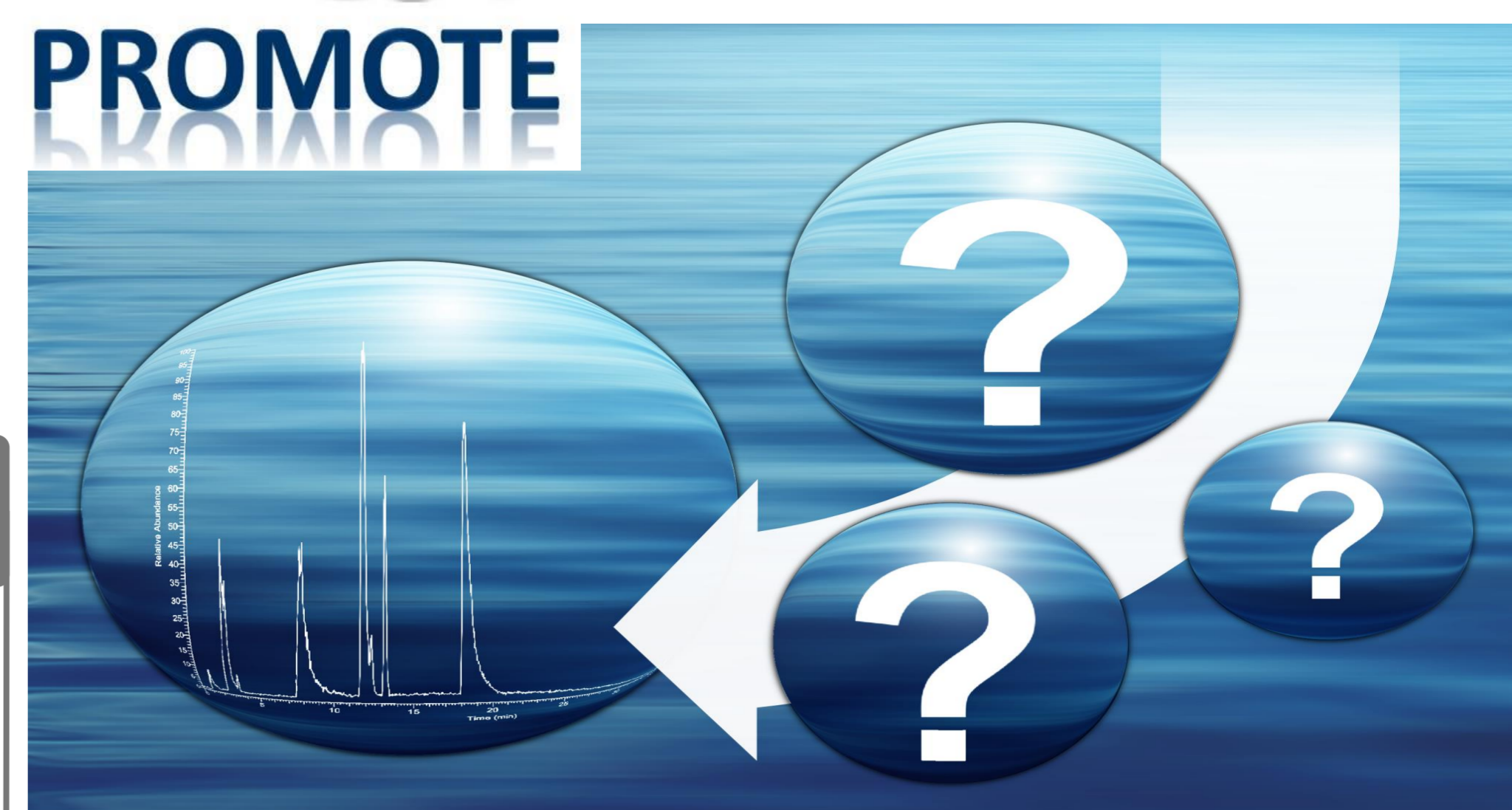


Non-target screening of highly polar organic micropollutants in environmental water samples

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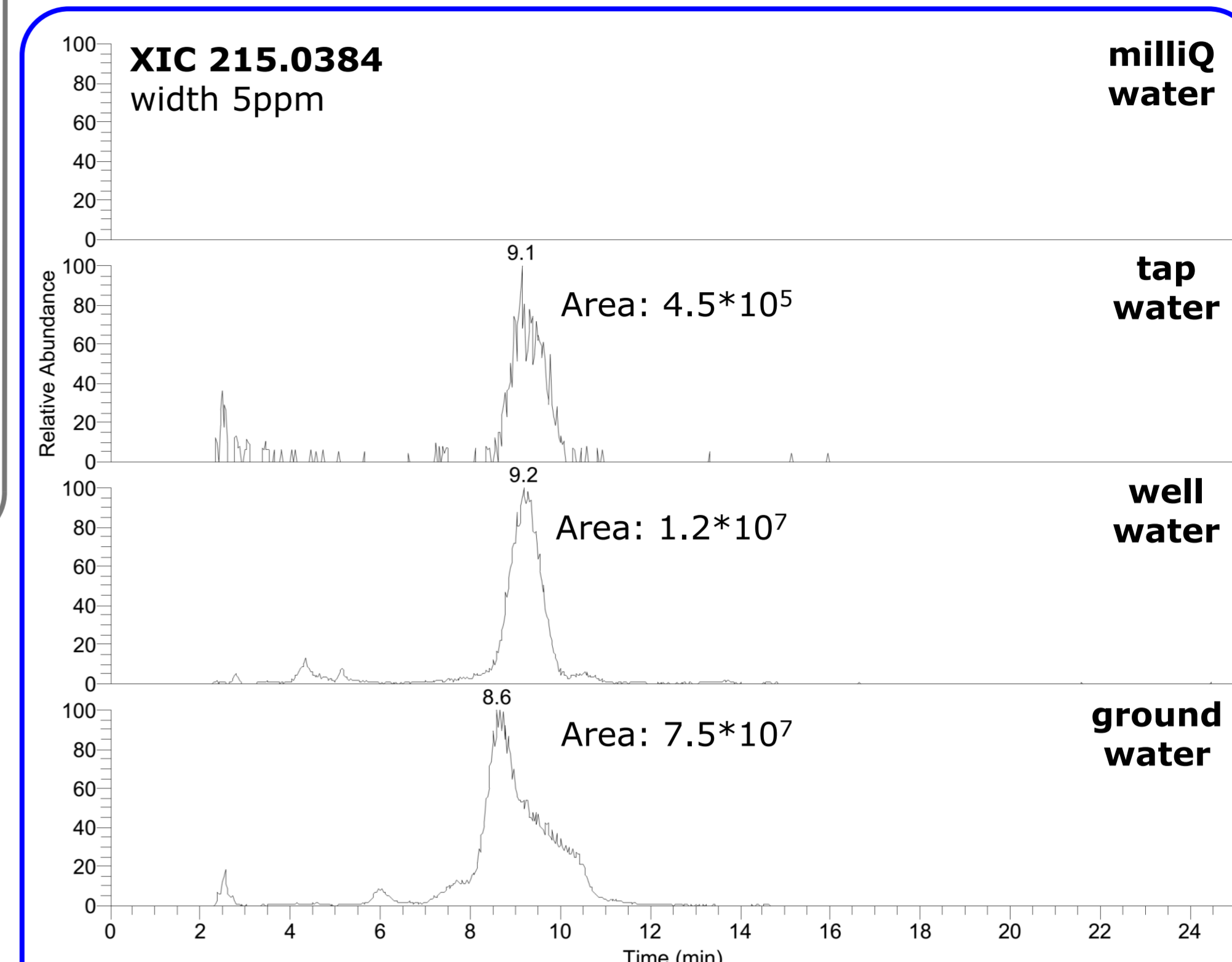
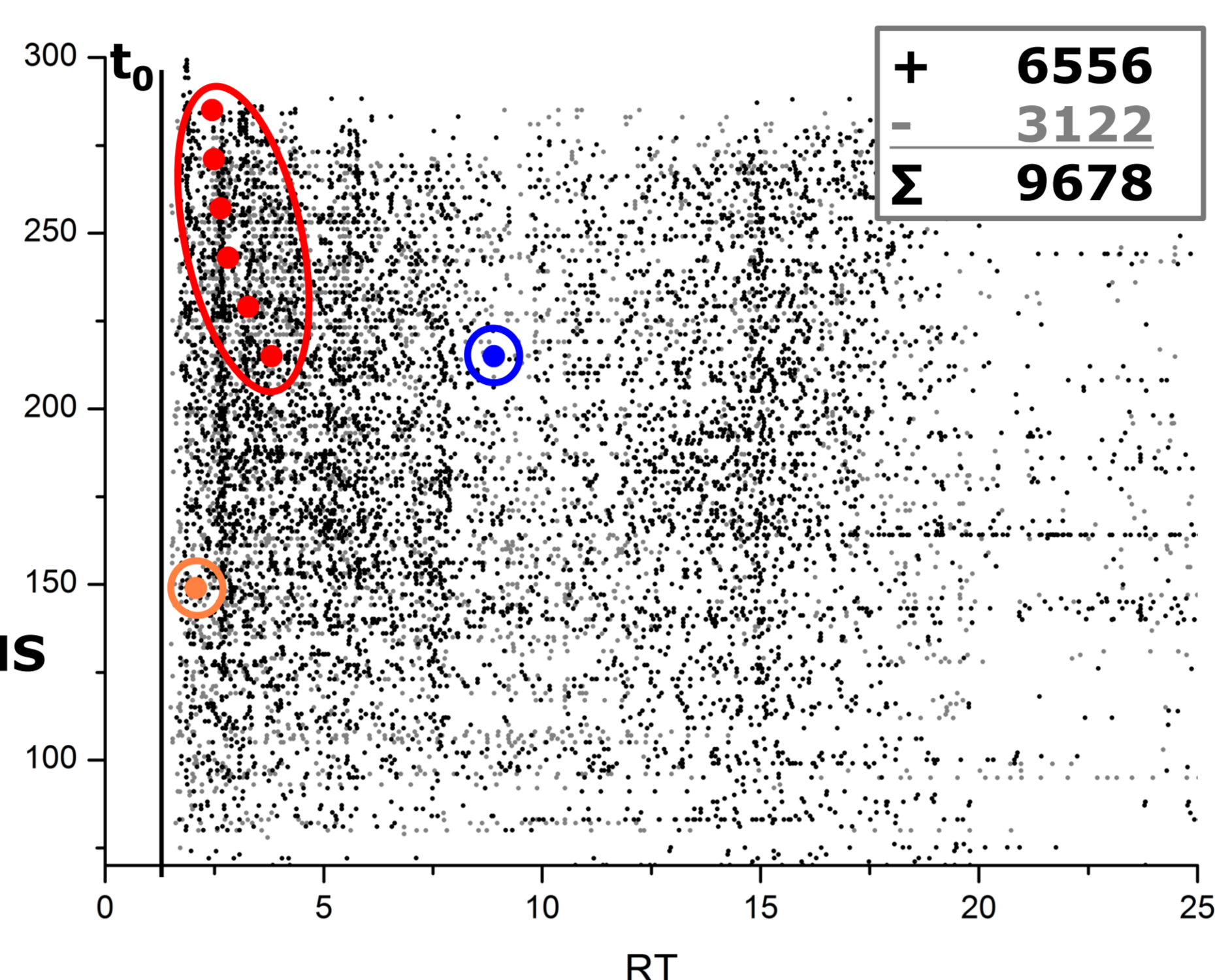
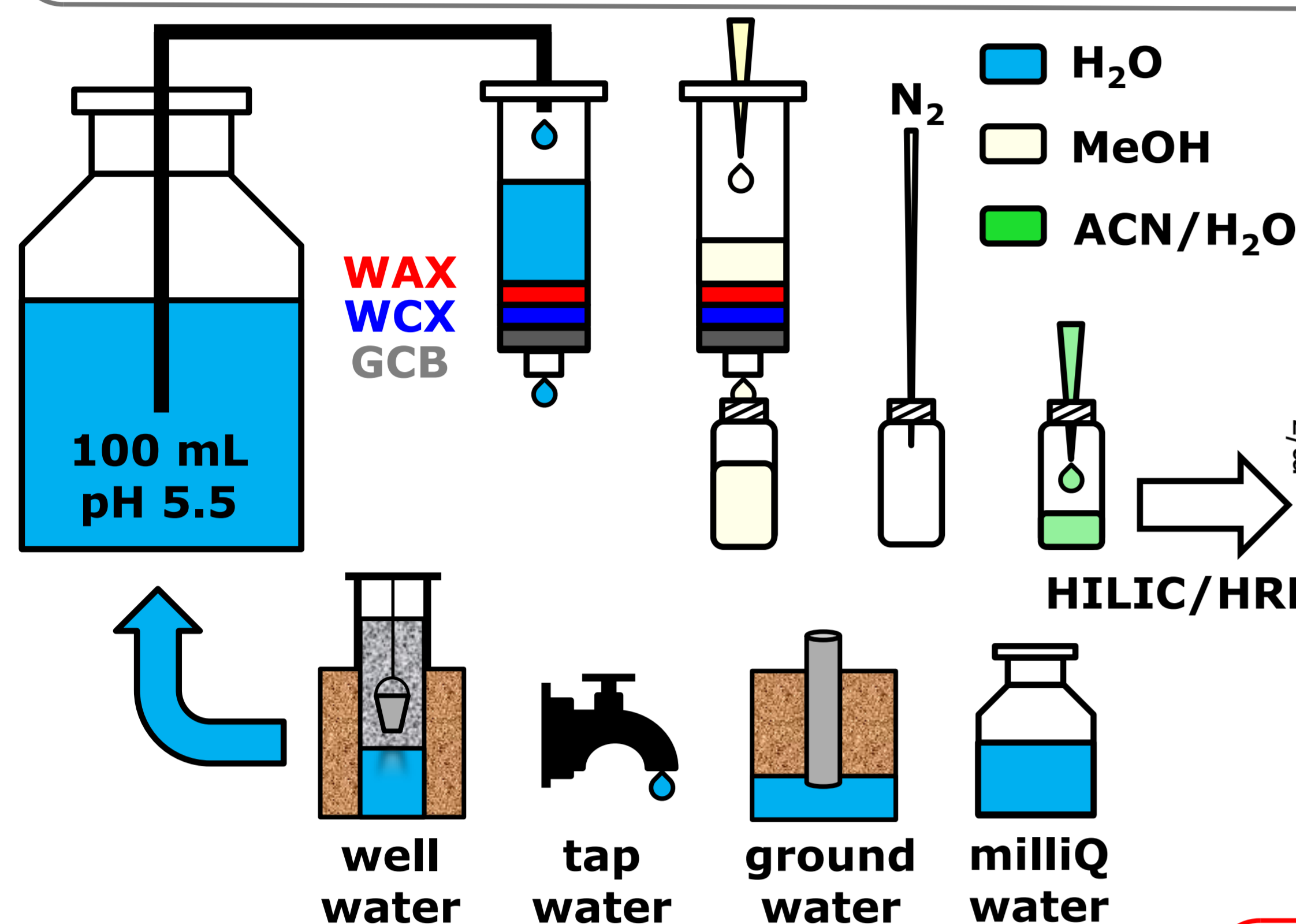


Introduction

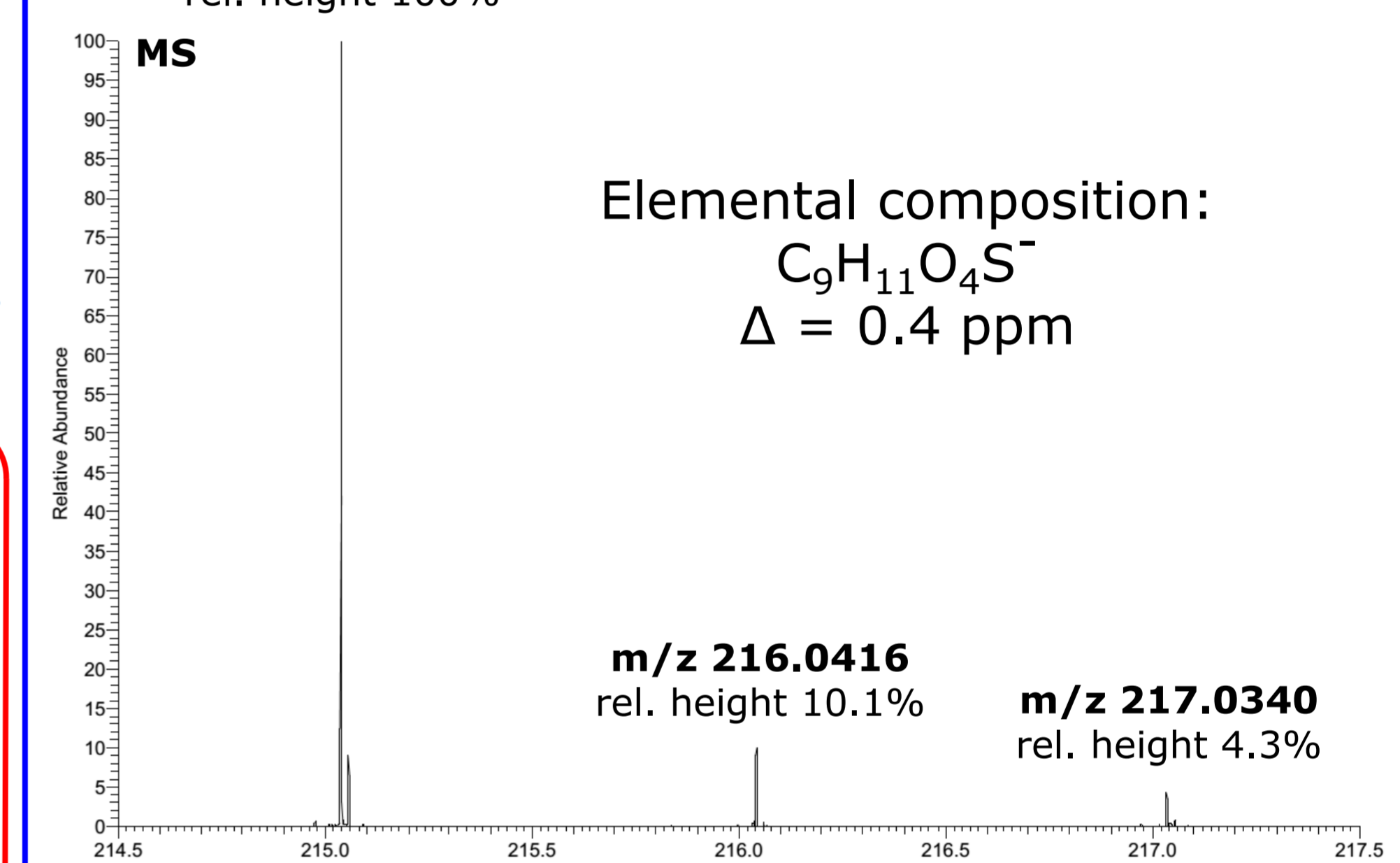
Highly polar organic substances may be able to penetrate natural and artificial barriers and are thus mobile in the water cycle. If these mobile organic contaminants (MOC) are persistent (PMOC) against microbiological and chemical degradation, their removal during waste water treatment and drinking water purification^[1] may prove difficult. If these substances are present in high concentrations, toxic or undergo toxication^[2], problems for the aquatic environment and human health may arise. As a consequence of the lack of established analytical methods for MOC, only limited

information about the occurrence fate and distribution of these substances is available. In the PROMOTE project, we strive to close this gap in knowledge.

In this work, a newly developed SPE^[3] method that allows enrichment of PMOC from aqueous samples is presented and deployed in combination with a suitable HILIC/HRMS method to perform a non-target screening for potential PMOC in environmental water samples, including polluted well and ground water, tap water and milliQ water as blank. Preliminary results of the non-target screening are shown.



m/z 215.0384
rel. height 100%

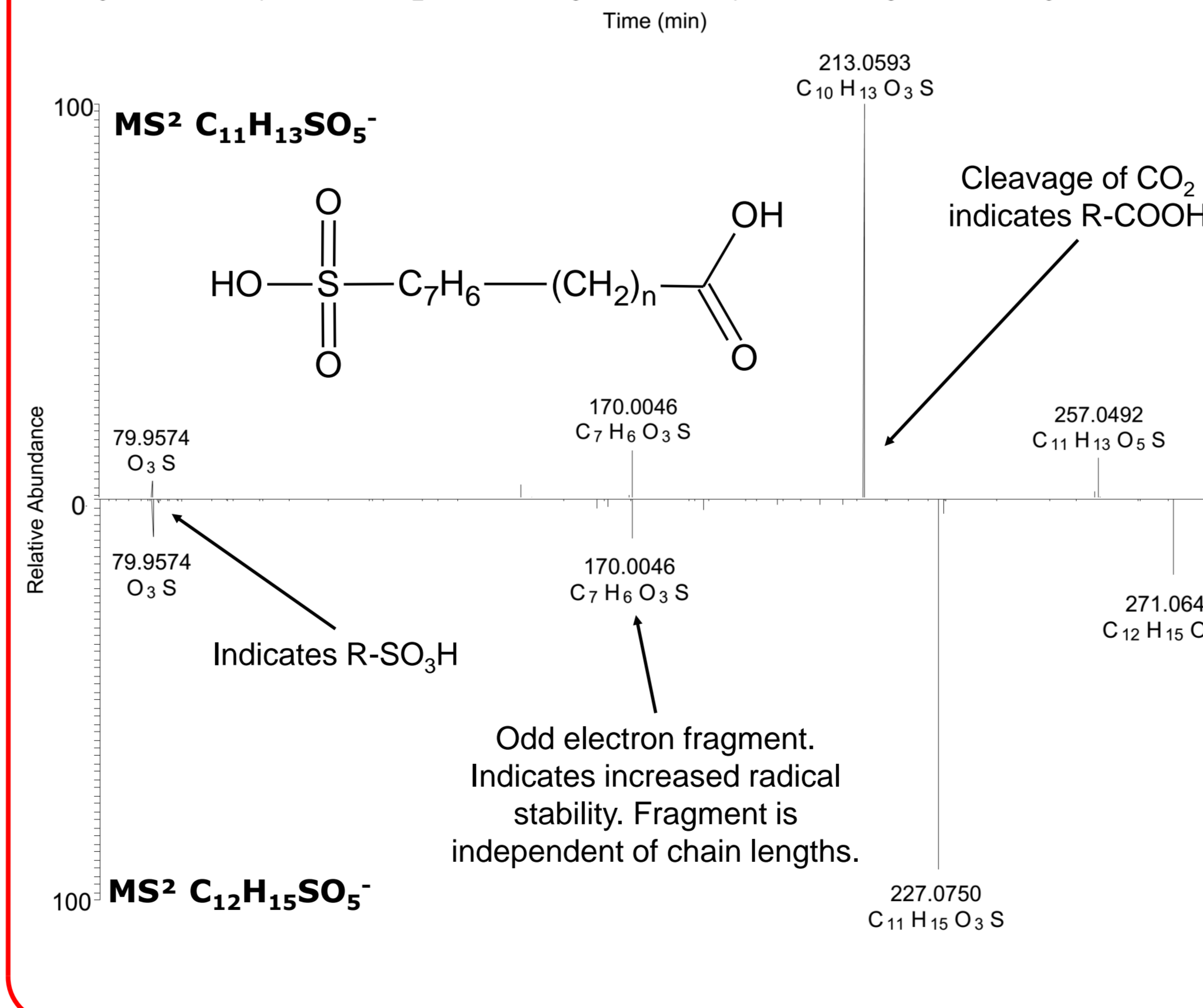
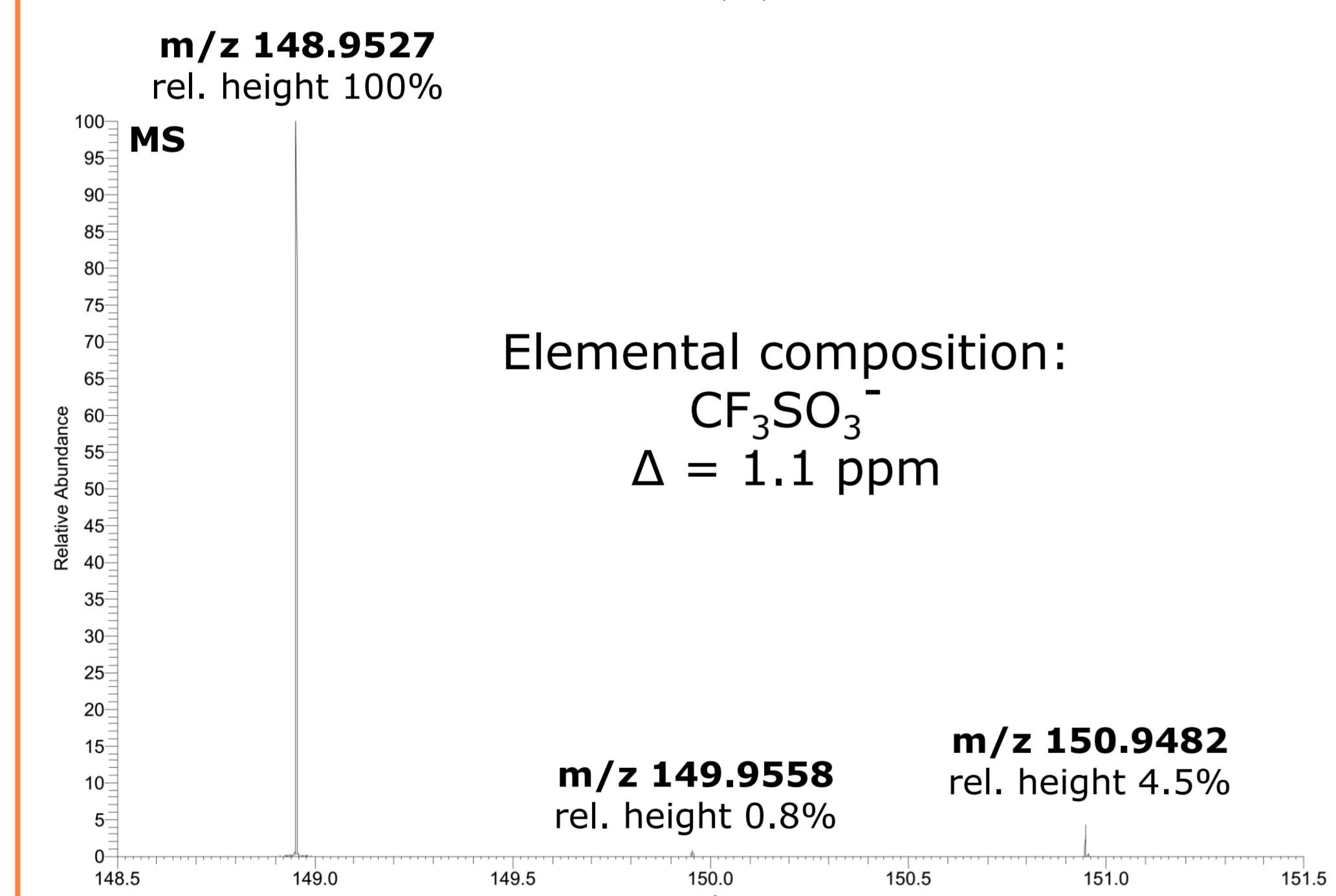
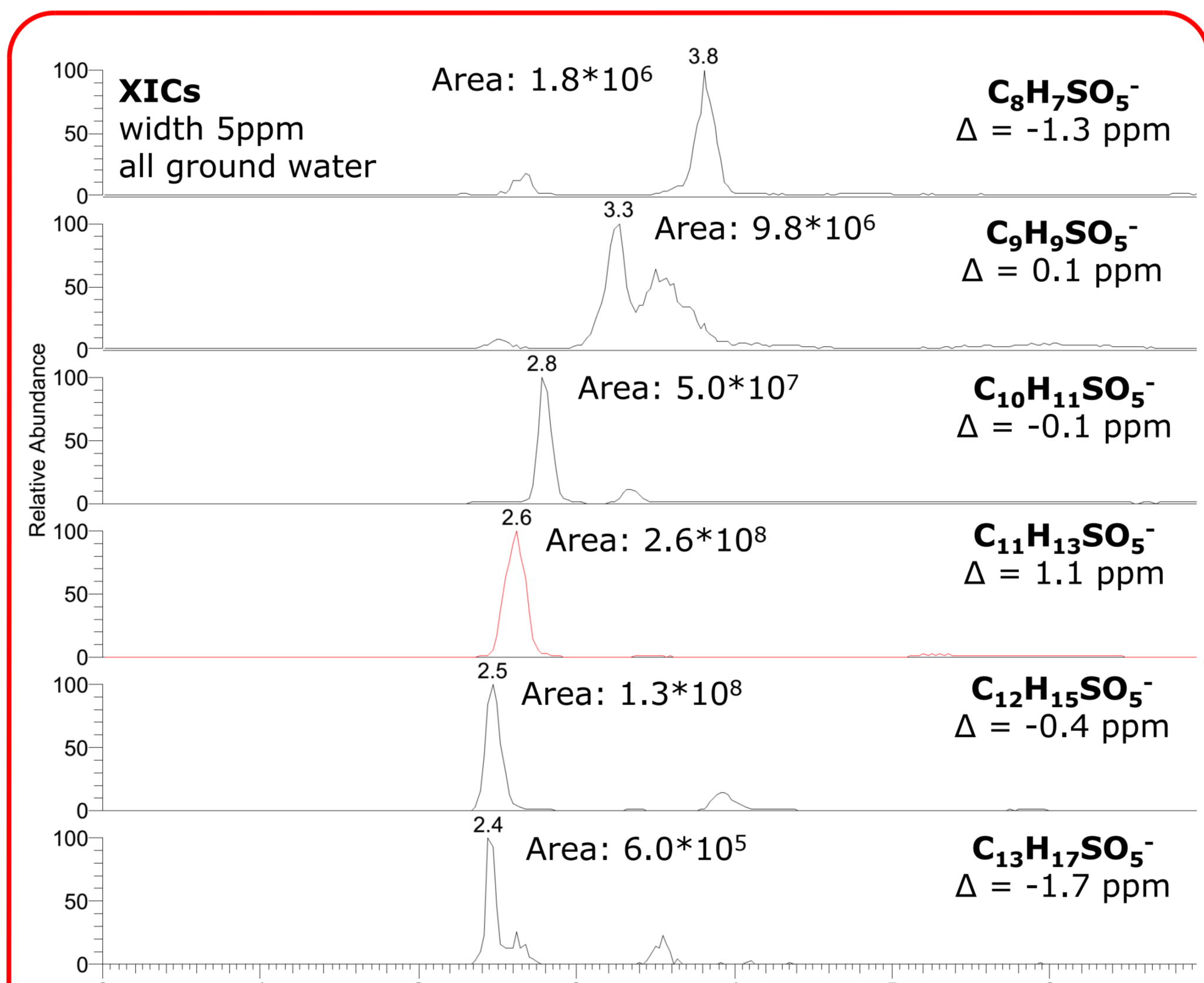
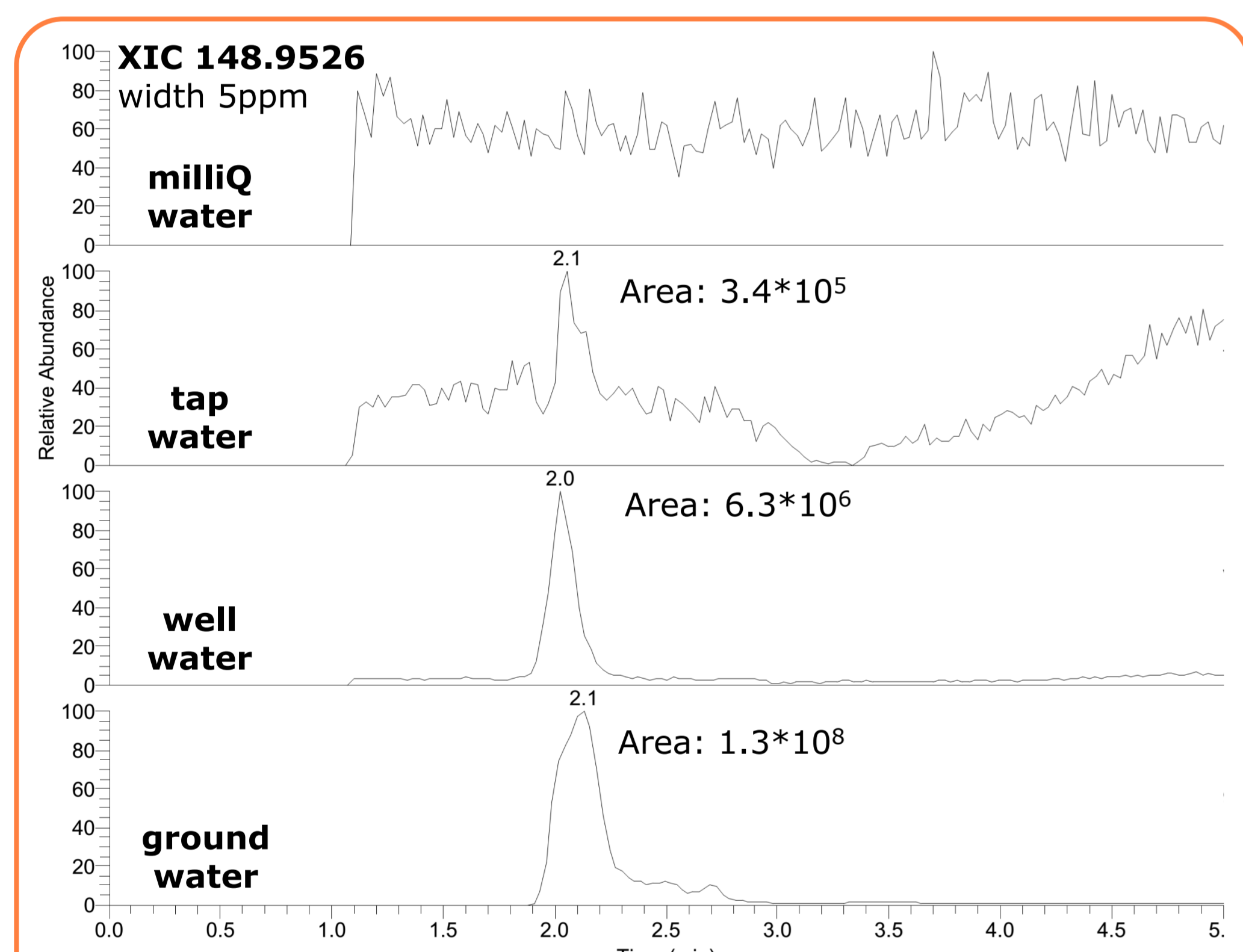


C₉H₁₁O₄S⁻ appears to have a sufficient persistency and mobility in the water cycle to reach drinking water, and is thus classified as PMOC. Structure and origin of C₉H₁₁O₄S are still under investigation.

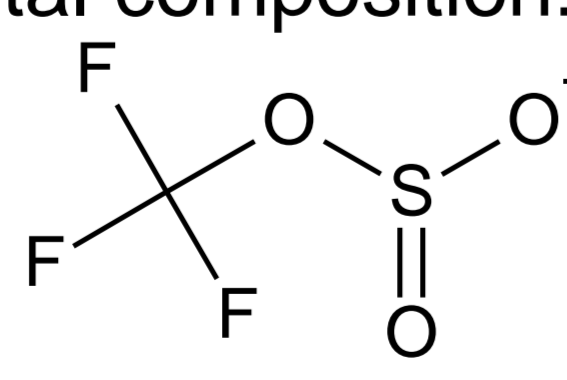
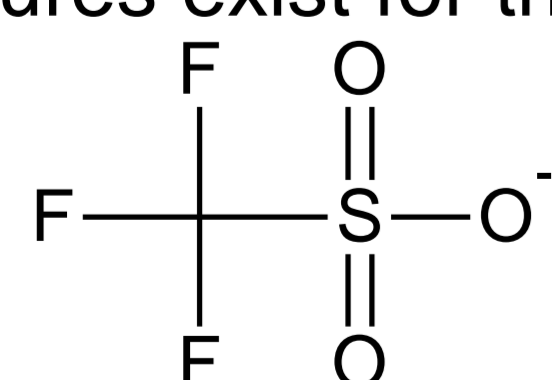
A supposed series of homologues was observed during the non-target screening. The MS² spectra of the most abundant homologues (assuming a similar ESI response) were compared to facilitate structural elucidation. In addition, fractions were collected, evaporated to dryness, reconstituted in D₂O/ACN 1:1 (v:v) and measured with HRMS by direct infusion.

For both homologues a cleavage of CO₂ as well as an SO₃⁻ fragment were observed, indicating the presence of a carboxylic acid and a sulfonic acid group respectively.

The fragment C₇H₆O₃S⁻ is identical for both homologues and violates the even electron rule, thus indicating a position of increased radical stability in the molecule. A benzylic position appears to be most likely but other alternatives cannot be ruled out by the available spectrometric information. The mass shift of 1 Da for the deprotonated molecule after H/D exchange is in compliance with the proposed structure. Structural elucidation of these substances is still ongoing.



The unequivocal elemental composition for the ion m/z 148.9526 is CF₃SO₃⁻. Only two chemically sound structures exist for this elemental composition:



Trifluoromethanesulfonate

Trifluoromethyl sulfite

Of the two proposed structures for m/z 148.9526 Trifluoromethanesulfonate, the anion of triflic acid, is more likely to occur in environmental water samples. Triflic acid has a high production volume (> 100 t/a) and is applied in various processes including vulcanization and polymerization. A confirmation of the proposed structure is still pending.

Conclusion and outlook

The preliminary screening results are a strong indicator that PMOC may impact drinking water. Identification of the detected substances, however is a laborious task and prioritization is required to select the most promising and important candidates for structural elucidation.

A more comprehensive screening campaign with a higher number of samples and a spatial spread over several European countries is in preparation.

References:

- [1] T.P. Knepper et al. *Waste Manage. (Oxford)* **1999**, 19, 77-99.
- [2] Schmidt, C. K.; Brauch, H.-J. *Environmental Science & Technology* **2008**, 42, 6340.
- [3] Zahn, D.; Frömel, T.; Knepper, T. P. *work in progress*

Acknowledgement:

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